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# Challenges and Opportunities when Managing Soil Sulfur

**2024 Indiana CCA Conference**

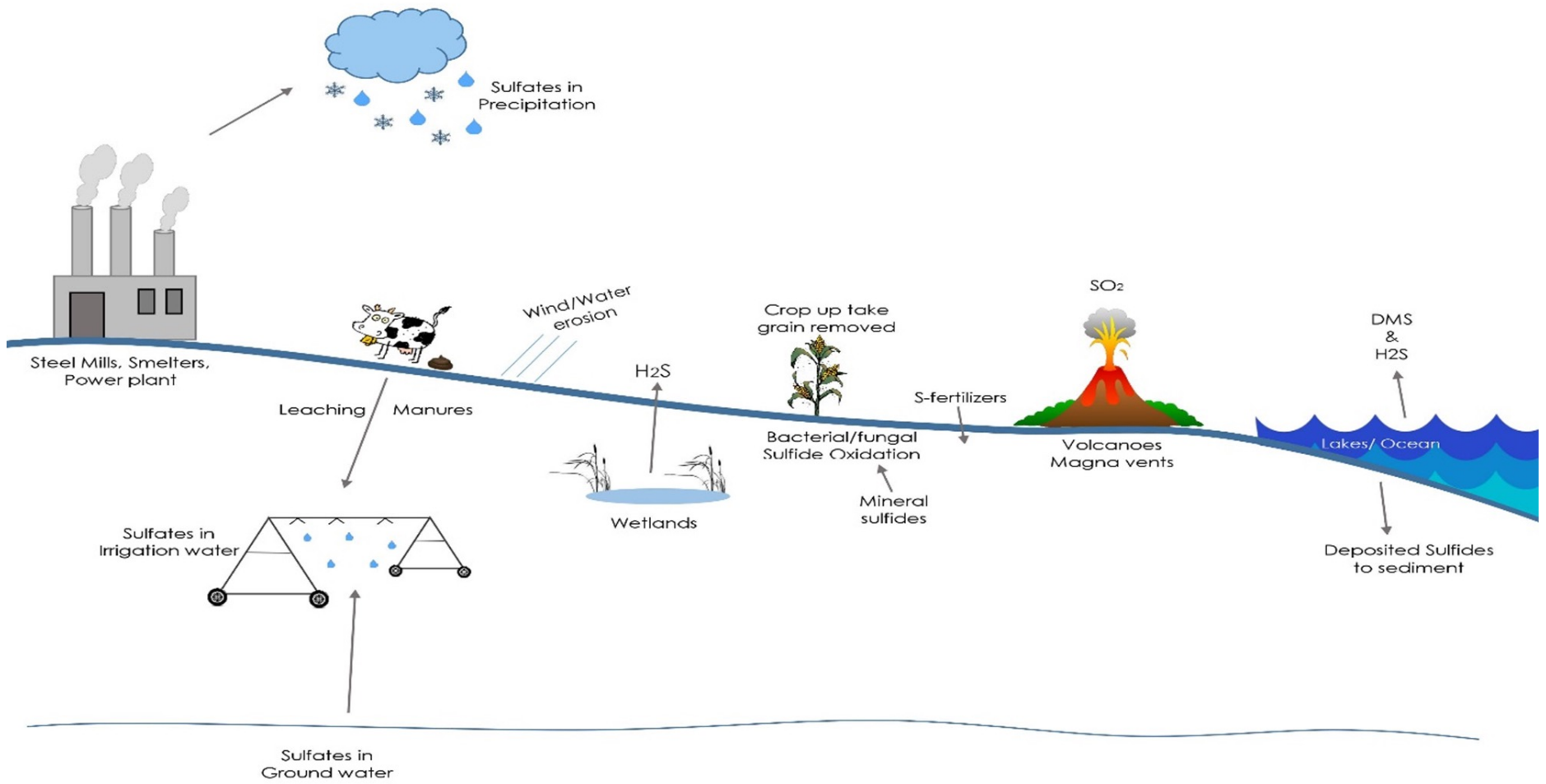
Dr. Brady Goettl

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# Topic to Cover

- S in our environment
- The role of S in crop growth
- Forms of S and S cycle
- Sources of S in agriculture
- Testing soil S levels
- Practical S-fertility management

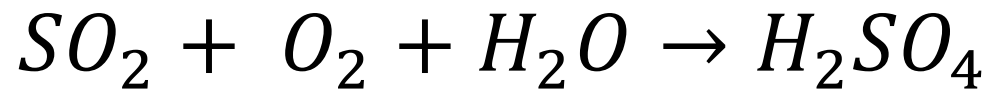




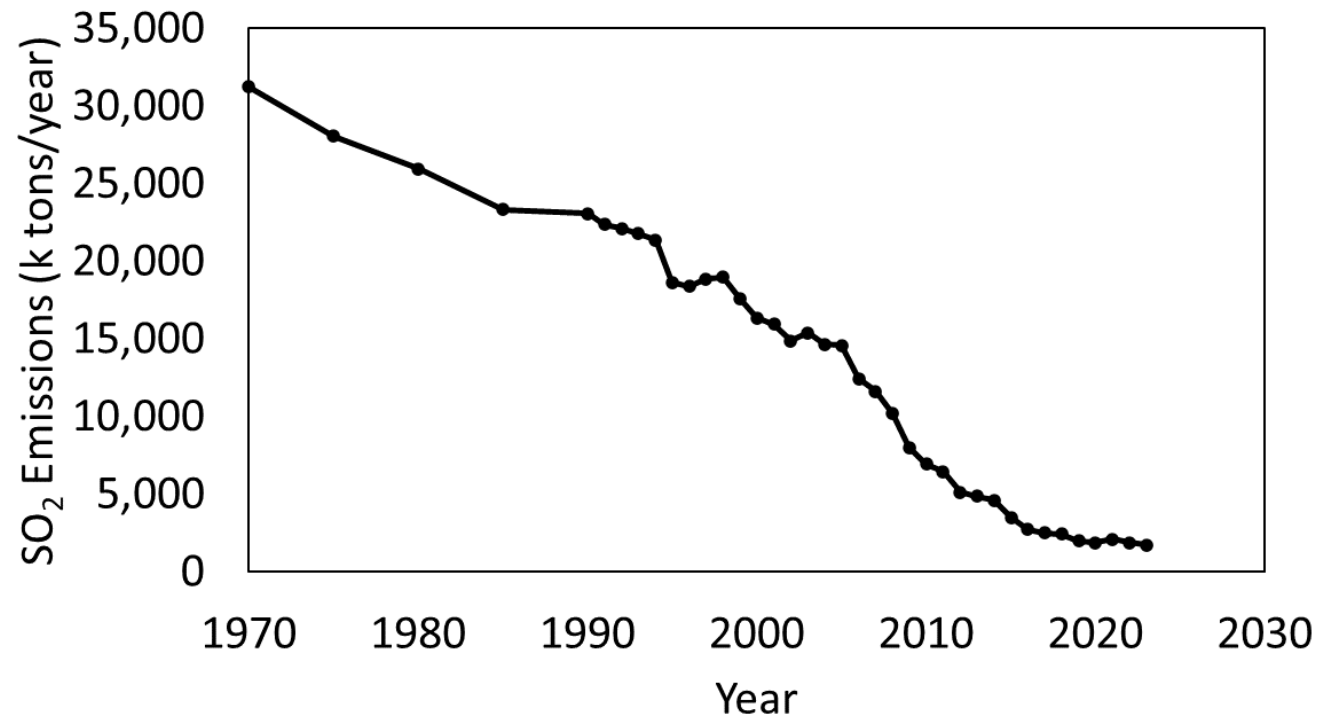
Source: D. Franzen, NDSU

# Sulfur in the Atmosphere

- Naturally released from wetlands, waterbodies, volcanic activity
- Product of coal and fossil fuel combustion

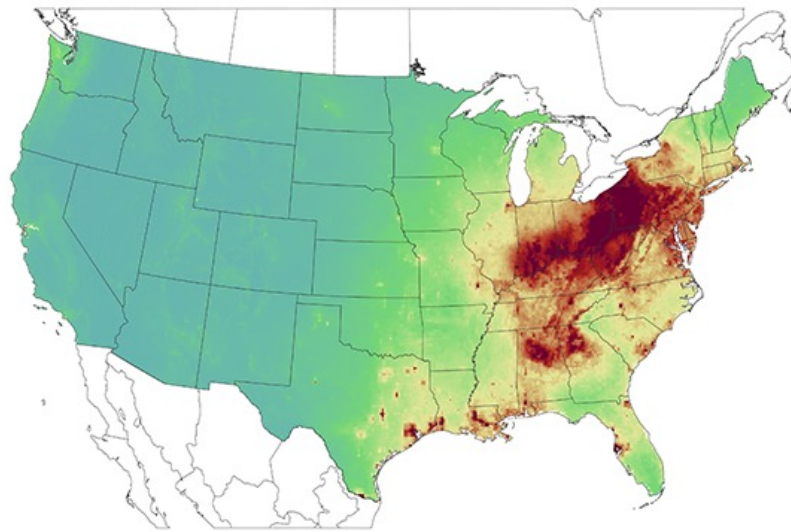


# US Anthropogenic SO<sub>2</sub> Emissions



## Three-Year Average of Total Sulfur Deposition

2000–2002

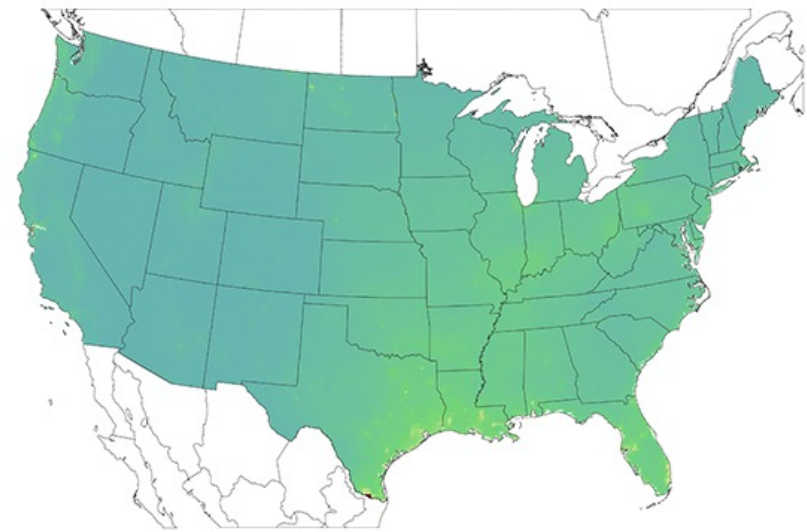


2018–2020

Total S  
(kg-S/ha)

A vertical color scale legend for sulfur deposition, labeled "Total S (kg-S/ha)". The scale ranges from 0 to >20, with increments of 2. The colors transition from dark green at 0, through light green, yellow, orange, and brown, to dark red at >20.

Total S (kg-S/ha)	Color
0	Dark Green
2	Light Green
4	Yellow-Green
6	Yellow
8	Light Orange
10	Orange
12	Light Brown
14	Orange-Brown
16	Dark Orange
18	Red-Orange
>20	Dark Red



Source: CASTNET/CMAQ/NADP  
USEPA, 2021

**“In 1860, sulfur was recognized as an essential nutrient... referred to as the ‘fourth major nutrient’ following nitrogen, phosphorous, and potassium”**



# The Role of Sulfur in Plants

- Component of amino acids—“the building blocks of protein”
- Component of *nitrate reductase*
  - Converts  $\text{NO}_3$  to organic N within the plant
  - Deficiency of S interferes with N metabolism

# Sulfur Deficiency

- Light green coloring of the whole plant
- Interveinal chlorosis
- Best diagnosed with tissue testing



# Is Sulfur Demand Increasing?

- Reduced SO<sub>2</sub> emissions
- Lower S concentration in synthetic fertilizers compared to organic sources
- Increased crop removal

# Sulfur in the Soil

## Inorganic

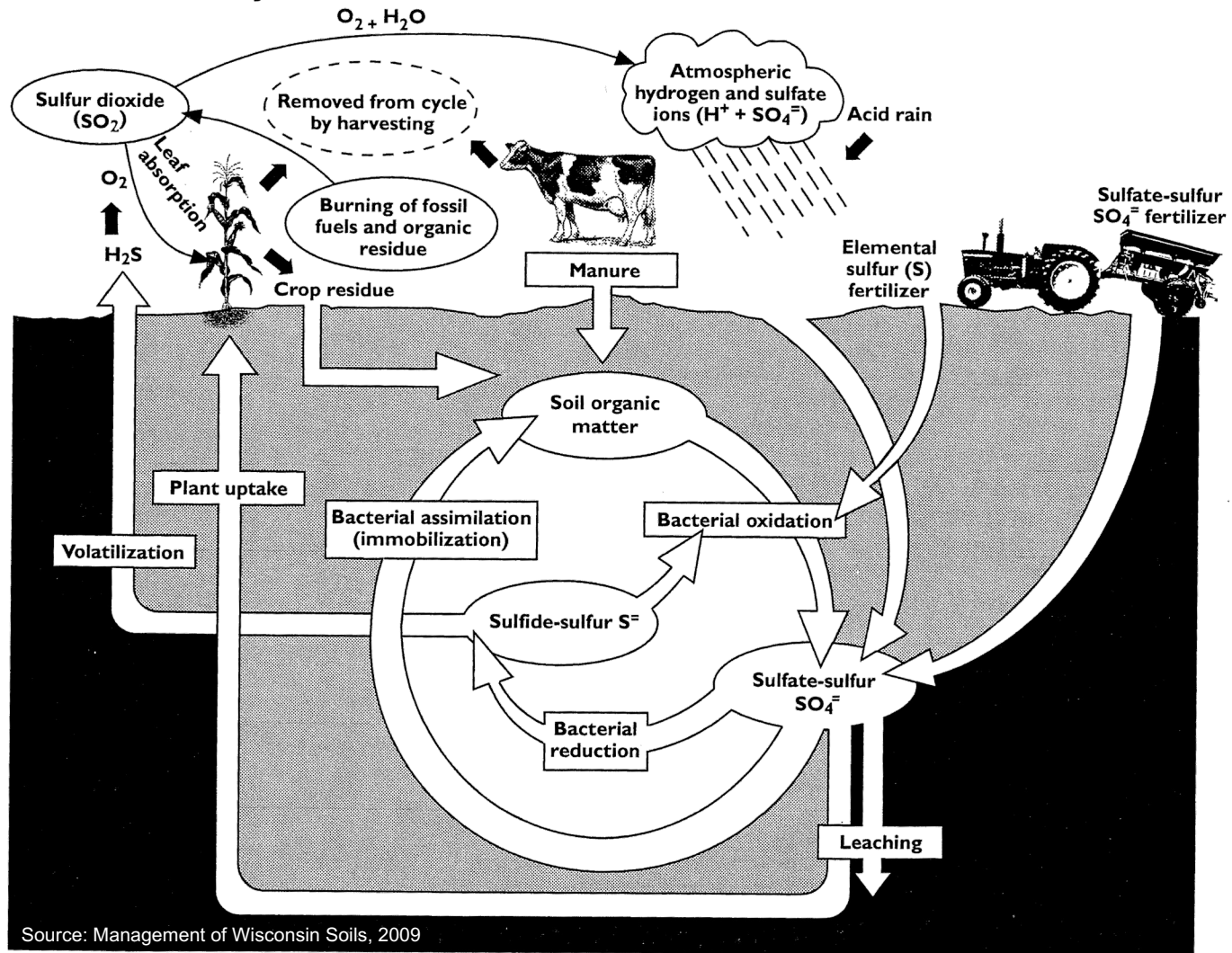
- Sulfate ( $\text{SO}_4^{-2}$ )
- Elemental sulfur (S)
- Hydrogen sulfide ( $\text{H}_2\text{S}$ )
- Salt/Mineral
  - Gypsum ( $\text{CaSO}_4$ )
  - Epsomite ( $\text{MgSO}_4$ )
  - Pyrite ( $\text{FeS}_2$ )

## Organic

- Organic sulfates
- Carbon-bonded sulfur

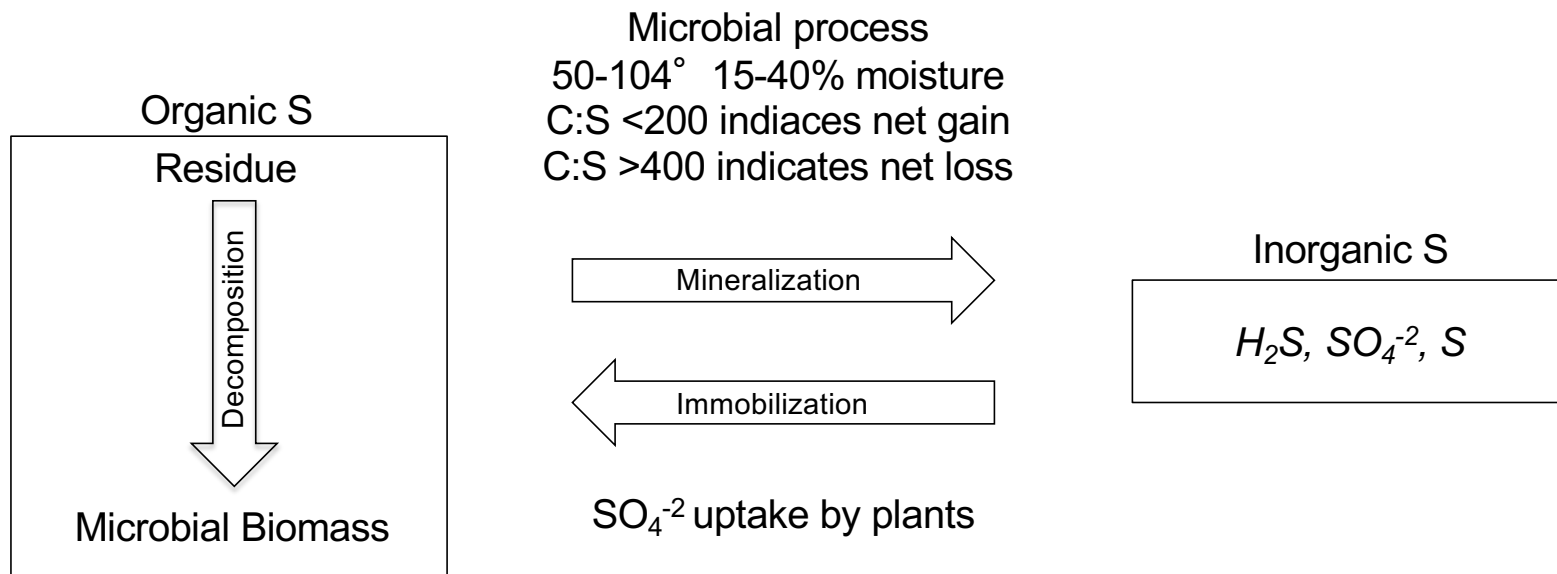
Most soils S is in organic forms, so S is highly correlated to organic matter

**Figure 9-6. The sulfur cycle.**

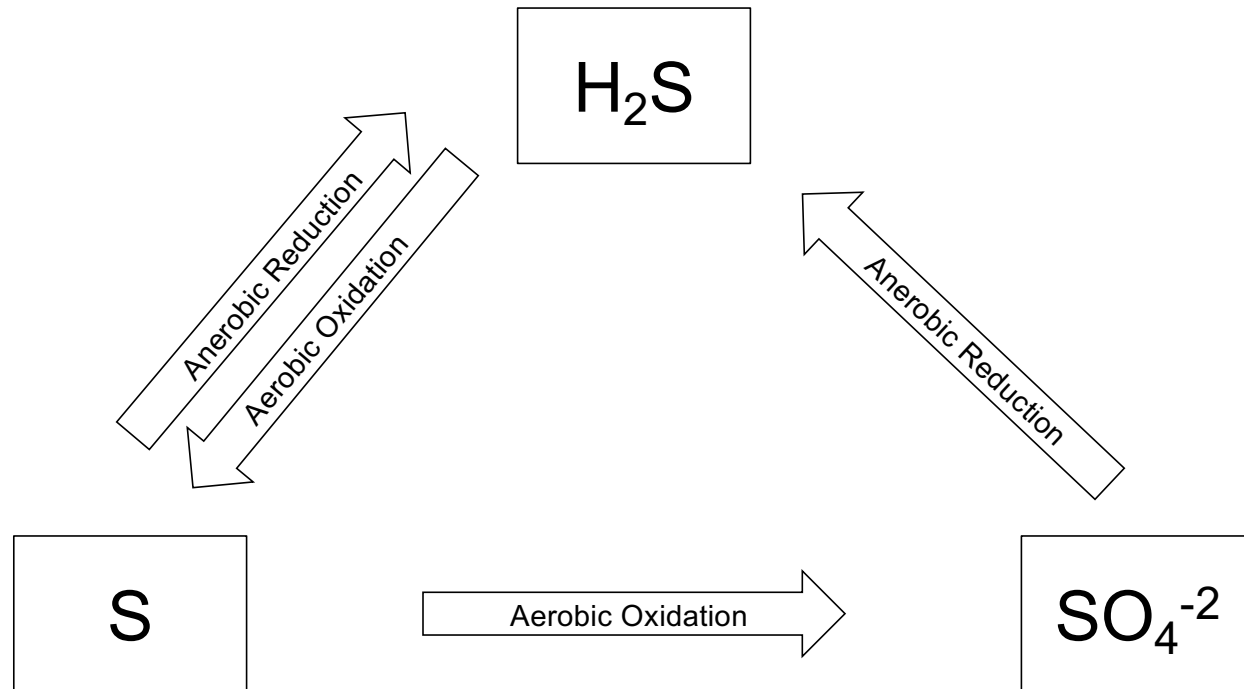


Source: Management of Wisconsin Soils, 2009

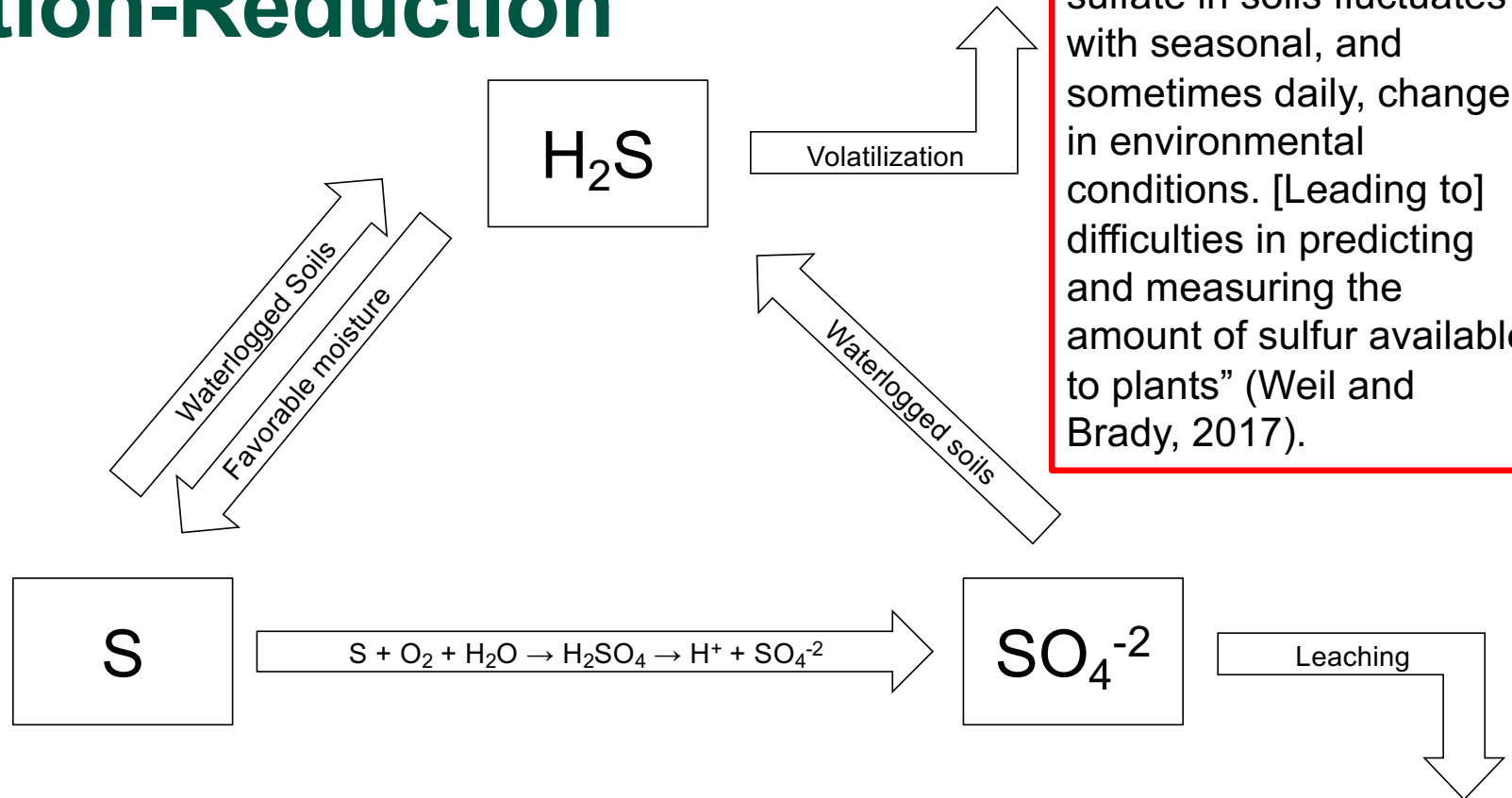
# Mineralization-Immobilization



# Oxidation-Reduction



# Oxidation-Reduction



“The supply of available sulfate in soils fluctuates with seasonal, and sometimes daily, changes in environmental conditions. [Leading to] difficulties in predicting and measuring the amount of sulfur available to plants” (Weil and Brady, 2017).



# Sources of Sulfur

- Organic (crop residue, manure)
  - Variable depending on crop and animal species
  - Approx. 60% of manure S is available the first year
- Fertilizer
  - Solubility varies
  - May require chemical transformations before it is plant-available

# Sources of Sulfur

**Table 9-12. Sulfur fertilizers.**

Material	Chemical formula	Fertilizer analysis N-P <sub>2</sub> O <sub>5</sub> -K <sub>2</sub> O	Sulfur content
		— % —	— % —
<b>Very soluble</b>			
Ammonium sulfate	(NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub>	21-0-0	24
Potassium sulfate	K <sub>2</sub> SO <sub>4</sub>	0-0-50	18
Potassium-magnesium sulfate	K <sub>2</sub> SO <sub>4</sub> •2MgSO <sub>4</sub>	0-0-22	23
Ammonium thiosulfate <sup>a</sup>	(NH <sub>4</sub> ) <sub>2</sub> S <sub>2</sub> O <sub>3</sub> + H <sub>2</sub> O	12-0-0	26
Magnesium sulfate (Epsom salts)	MgSO <sub>4</sub> •7H <sub>2</sub> O	0-0-0	14
Ordinary superphosphate	Ca(H <sub>2</sub> PO <sub>4</sub> ) <sub>2</sub> + CaSO <sub>4</sub>	0-20-0	14
<b>Slightly soluble</b>			
Calcium sulfate (gypsum)	CaSO <sub>4</sub> •2H <sub>2</sub> O	0-0-0	17
<b>Insoluble</b>			
Elemental sulfur	S	0-0-0	88–98

Similar effectiveness whether surface applied or incorporated

<sup>a</sup>Ammonium thiosulfate is a 60% aqueous solution. Source: Management of Wisconsin Soils, 2009

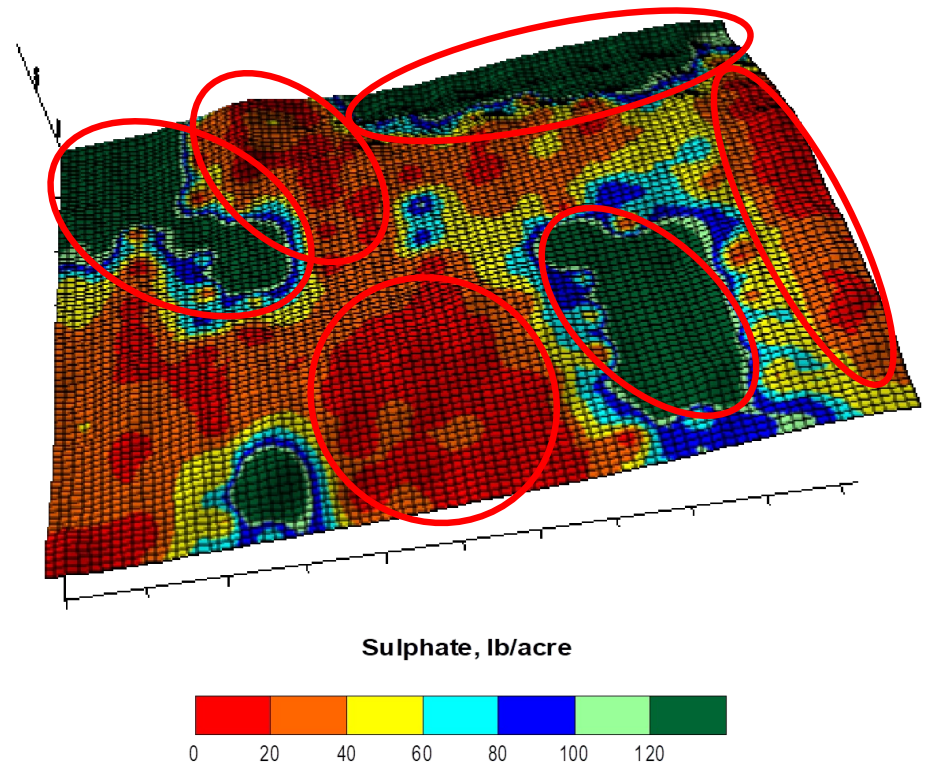
# Elemental Sulfur

- Very insoluble
- Requires oxidation before it is plant available
- Oxidation rate depends on:
  - Particle size
  - Soil mixing/incorporation
  - Soil moisture and temperature
  - Microbial activity
- Sulfur deficient fields should include  $\text{SO}_4\text{-S}$  application for immediate availability



# Sulfur Across the Landscape

- Gypsum deposits in the North Central Region
- Groundwater in ND, SD, and Western MN is enriched with gypsum and other sulfate salts
- Groundwater moves sulfate to the surface



# Sulfur Response in ND

- Severe S deficiency became evident in early 1990s with the introduction of canola

S rate (lb/ac)	S source	Canola Yield (lb/ac)		
		Hilltop	Slope	Footslope
0		30	240	1460
20	AMS	1650	1670	1720
40	AMS	1800	1860	2170
40	Elemental	620	1060	1630

6,000% Yield Increase!

# Developing a Soil Test

- Correlation
  - Relationship between plant uptake of a nutrient and the amount of nutrient extracted by a soil analysis
  - Test should quantify the amount and form of nutrients taken up by the plant



# Developing a Soil Test

- Calibration
  - Establish meaning between soil test measurements and crop response across a wide array of fields and fertility levels



# Sulfur Soil Tests

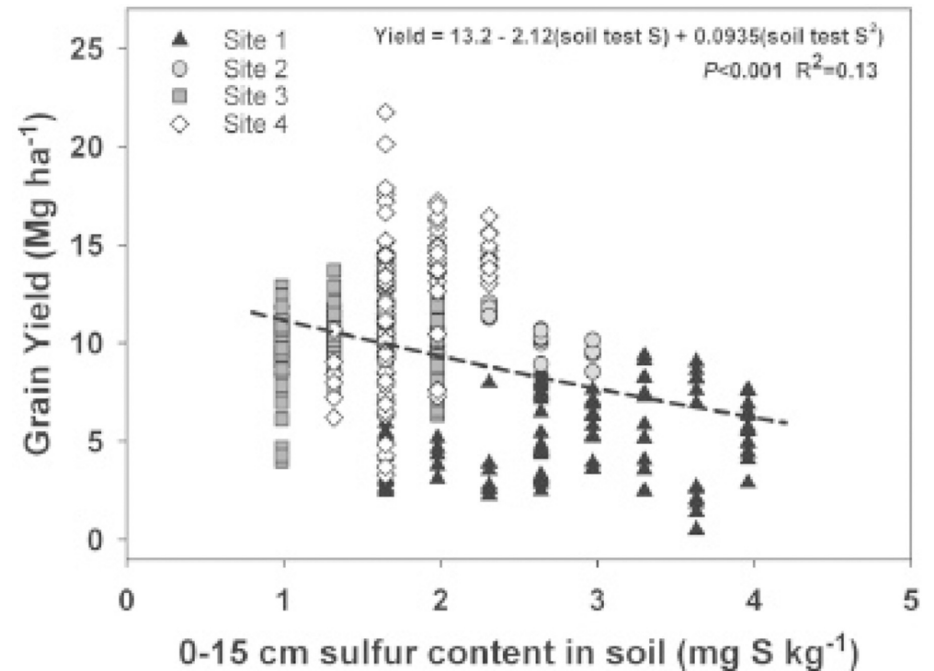
- No fully satisfactory  $\text{SO}_4^{-2}$  tests exist
  - S is constantly undergoing transformations
  - Tests do not reflect sources of potentially available S
  - High mobility of  $\text{SO}_4^{-2}$
  - Precipitated gypsum in arid/semi-arid environments





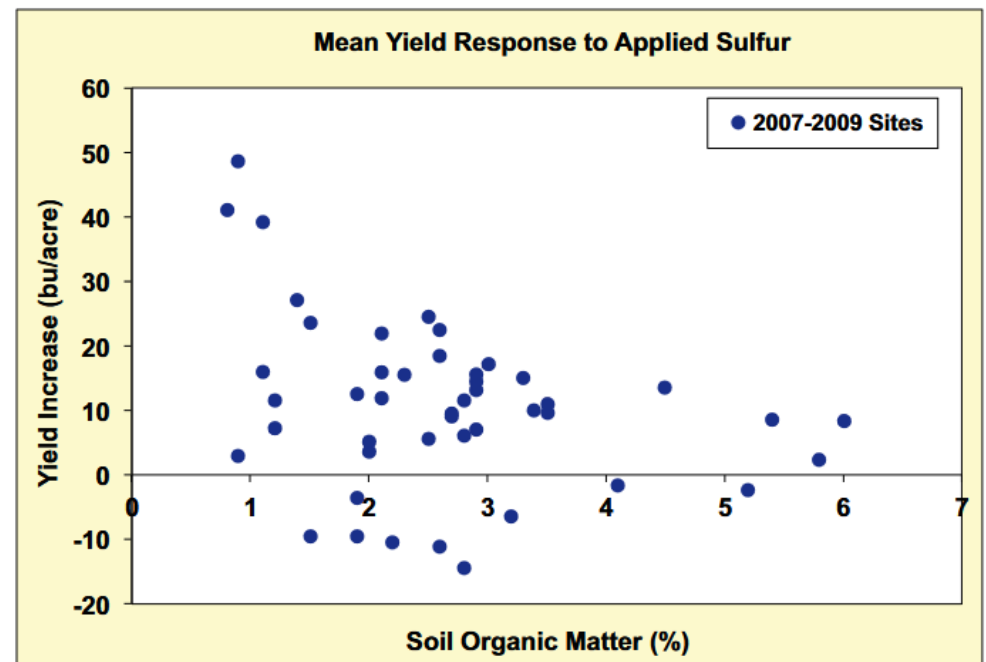
# Sulfur Soil Test Calibration (or Lack of)

- Minnesota, 2008-09
- S increased corn yield at two of four locations
- Greatest response to S occurred in soils with <2% OM
- Response was not related to SO<sub>4</sub>-S soil test



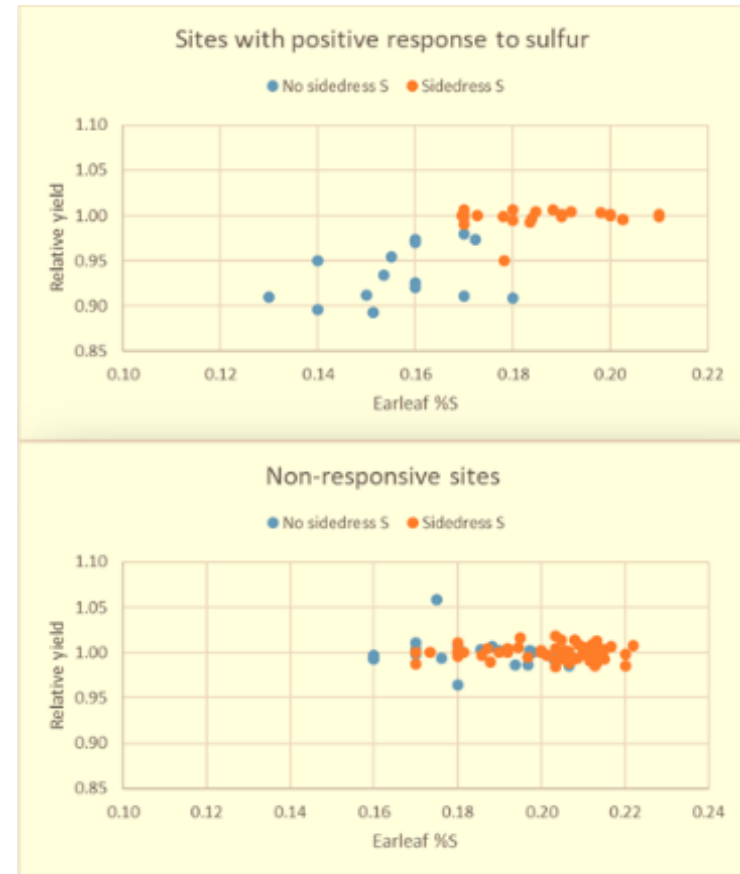
# Sulfur Soil Test Calibration (or Lack of)

- Iowa 2007-2009
- S fertilizer increased yield at 28 of 47 sites
- 0-6 in. depth  $\text{SO}_4\text{-S}$  soil tests were not related to yield response
- Response to S was greater in low OM soils



# Sulfur Tissue Test

- Indiana, 2017-2022
- Corn yield was increased by sidedress S in 15 of 40 trials
- Pre-sidedress S concentration  $< 0.18\%$  may benefit from S application



# Practical Sulfur Management

- Do all fields/crops require a sulfur application?
  - Likely not
- How do we know when to apply S when the tests are not good predictors?
  - Consider aggravating factors of S deficiency

# Practical Sulfur Management

- Pay particular attention to crops with high S demand
  - Alfalfa, canola/forage brassica, cole crops, high biomass crops
- Conditions favorable to S deficiency:
  - High precipitation, low soil OM, coarse soil textures, high water table, saturated conditions

# 4Rs of Sulfur Management

- Right *rate*
  - Recommendations range from 10-50 lb S/ac
- Right *source*
  - $\text{SO}_4\text{-S}$  is plant available, elemental S must be oxidized
- Right *time*
  - Spring applied  $\text{SO}_4\text{-S}$  has a lower risk of leaching loss
- Right *placement*
  - Broadcast, incorporated, or banded are equally effective

# Take-Home Points

- S deficiency is becoming more common
- Diagnostic tests are not reliable for determining crop demand/response to S
- Soil and field conditions can be help indicate the potential need for fertilizer S
- Sulfate fertilizers should be used to ensure plant availability



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We find it just as scientific  
to be practical as it is  
practical to be scientific.

**Hugh Hammond Bennett**